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Optical and electrical properties of reactively sputtered TiN, ZrN, and HfN thin films

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Thin films of titanium, zirconium and hafnium nitride are prepared by DC magnetron reactive sputtering at room temperature on fused silica, optical glass and silicon substrates. Deposition parameters are investigated in order to obtain stoichiometric films, The optical and electrical properties of the films as a function of nitrogen partial pressure are determined. The results show that an inverse correlation exists between the optical reflectance and the electrical resistivity of the films. The optical constants of the films are determined by Variable Angle Spectroscopic Ellipsometry (VASE) measurements from 240--1700 nm at 10 nm steps. Deposited film composition is obtained by the Rutherford Ion Back Scattering (RBS) method. The rms roughness of the films is measured by using an optical scatterometer. Ellipsometer data for all three films show that their refractive index (n) in the visible spectrum is decreased by increasing the film thickness while the extinction coefficient (k) is unchanged. Thin films of TiN have the lowest room temperature resistivity ($\cong 75 \, \mu\Omega$ - cm) relative to ZrN and 1 lfN thin films.

Keywords: Reactive sputtering, optical coatings, optical properties, electrical properties, thin films, transition metal nitrides.

1. INTRODUCTION

Titanium nitride ('l'iN), zirconium nitride (ZrN) and hafnium nitride (1 lfN) have optical, electrical and mechanical properties that are advantageous for applications to optics, electronics and tribology. Generally, nitride films have been deposited by chemical vapor deposition (CVD) or physical vapor deposition (PVD). However, CVD involves high temperature processing and PVD produces porous films with poor adhesion. Two other deposition methods have also been employed for making nitride coatings: magnetron reactive sputtering and electron cyclotron resonance (1 CR) deposition. In recent years magnetron sputtering in recent years has become the leading coating process, compared to CVD and J'VD because of these advantages:

- Magnetron sputtering can be done from Iarge-area targets with different geometry, which simplifies the **problem** of coating of large area substrates with uniform film thickness;
- The composition of sputter-deposited films can be tightly controlled;
- The device damage from X-rays generated during electron beam evaporation is eliminated;
- Coatings can be accomplished at relatively higher deposition rates; and
- Fully dense films are produced at room temperature,

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The aim of this work was to produce nitride films with desirable optical, electrical and mechanical properties that can survive harsh environments. The most important objective was to develop a reactive magnetron sputter deposition process at room temperature for TiN, ZrN, and 1 lfN films that yields a coating with the following properties:

- Reflectance from the UV to far-n{ optical range with values near theoretical maximum;
- Good adherence to substrates, particularly fused silica, optical glass and silicon;
- Low scatter films; and
- I lard coatings.

2. SAMPLE I'R11I'ARATION

The DC magnetron sputtering system we set up meets our requirements for stability, good uniformity, ability to produce multi layer films, and high production rates. The system is à 76 x 76 x 61-cm box coater with a mechanical roughing pump and a 30 cm-diameter high-vacuum cryopump with a water vapor pumping speed of 10,000 liters per second, This high pumping speed prevents reactive gases from poisoning the target material and helps achieve stability. The system's lowest attainable base pressure is 3 x 10-8 torr. Two 15 cm- and one 8 cm-diameter magnetron cathodes are mounted at the bottom of the deposition chamber, meeting the demand for producing multilayer films with good uniformity. This arrangement makes it possible to deposit nitrides, oxides, and compounds of three materials without venting the chamber. Up to three gases can be flowed into the chamber at any given time. The reactive and working gases used in the process are ultra-high-purity grade 5. Flow rates of the gases into the chamber are controlled separately by precision mass flow valves permitting Ar:N2, Ar:02 and N2:02 ratios to be varied at will. The system is equipped with both an optical and a quartz crystal monitor, enabling control of the deposition rate by either method.

All three starling materials are metal targets with 99.95 ?/o purity. The substrates are not intentionally heated or biased during deposition, 10 the reactive sputtering process an argon flow of 25 seem (cubic centimeter per minute at S'J'J') and a nitrogen gas flow of 3, 5, 10, 15, 20, 25, or 30 seem are used, resulting in a deposition pressure in the range of 2.8- 5×10^{-3} torr. Coatings are deposited onto optical glass, fused silica, and silicon substrates. The deposition rate for all t hree coal ings is $3-5\text{\AA/sec}$ with a source-to-substrate distance of 25 cm. All deposit ions started at a base pressure of 6×10^{-7} torr.

3. DETERMINATION OF OPTICAL CONSTANTS USING ELLIPSOMETRY

3.1 Samples

For each material, four samples are prepared on fused silica substrates. Two of the films are intended to be optically thick (opaque), while the other two were much thinner at 30 40 nm. For each type (thick and thin), one of the substrates was roughened on the back side, while the other was left smooth.

3.1.2 Measurements

Variable angle spectroscopic ellipsometry (VASE) measurements Were made on all samples from 240 nm to 1700 nm at 10 nm steps at external angles of incidence (ϕ_0) of 70, 75, and 80° using a commercial rotating analyzer-type ellipsometer. Some small artificial oscillations occur in VASE data at long wavelengths due to imperfections in the polarizers. As a result, small oscillations also occur in the derived n and k data at long wavelengths. No attempt was made to smooth the data.

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3.2 Analysis

3.2.1 Thick samples

Transmittance data confirmed that the thicker films were opaque (f $^{\circ}$ O over the ent ire 240--1700 nm spectral range). 1'bus, these films could be treated as substrates since the light did not penetrate to the fused silica interface. For a simple substrate, the ellipsometric data can be converted directly into a pseudodielectric function ε using

$$\varepsilon = \sin^2 \phi_0 \left[1 + \left(\frac{\dot{1} - \rho}{\dot{1} + \rho} \right)^2 \tan^2 \phi_0 \right], \tag{1}$$

where $\rho(\lambda, \phi_0)$ is the measured complex ellipsometric parameter, usually given in terms of two angles, Ψ and Λ .

$$p = \tan \Psi e^{i\Lambda}. \tag{2}$$

To the extent that the surface is perfectly smooth and has no overlays, as assumed here, the pseudodielectric function and the true dielectric function are ident ital. Corresponding pseudo-values for n and k are found from:

$$\varepsilon = \varepsilon_1 + i\varepsilon_2 = \dot{N}^2 = (n + ik)^2. \tag{3}$$

Pseudo-values from different angles ϕ_0 are identical (except from random noise), so the average over all angles was taken.

3.2.2 Thin samples

In this case the exact thickness (t) and n and k at each wavelength (λ) are unknown. VASE data were analyzed using a one-film model in which the fused silica substrate's optical constants are known, and initial guesses for t, $n(\lambda)$, and $k(\lambda)$ of the film are given. The ellipsometric parameter p is calculated from:

$$\rho = \frac{R_{\rho}}{R_{c}},\tag{4}$$

where $R_p(R_s)$ is the complex reflection coefficient for light polarized parallel (perpendicular) to the plane of incidence, These were calculated from the one film model using standard formulas. The values of t, $n(\lambda)$, and $k(\lambda)$ were varied to get a best-fit to the data, using a χ^2 minimization algorithm.

Correlation between t and the film's optical constants is a typical problem in the fitting. That is, there is no unique set of t, n anti k that produces the same or nearly the same best fit to VASE data. Simultaneously fitting both VASE anti 7' data removes the correlation. (This was the reason for preparing the thin films on transparent substrates.) The approach worked well. We obtained good tit for both VASE and 7' spectra with unique best-fit, t, anti t and t values, As an independent check, t and t data were used to fit the VASE data from the thin layer samples with roughened' substrate backsides using only t as a fitting parameter. We obtained very good fits, indicating that t and t fitted values were reasonably accurate.

3.2.3 Lorentz oscillator model

in addition to fitting n and k individually at each wavelength, a simple Lorentz oscillator model was used to fit n and k with only a few wavelength-indei>cndent parameters. In this model, ε of the film (either thick or thin) was modeled as:

$$\varepsilon(E) = \varepsilon_{\infty} - 1 \sum_{i=1}^{n} \frac{A_{i}}{E^{2} \cdot E_{i}^{2} + j \Gamma_{i} E}, \qquad (5)$$

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where E=hv is the photon energy, and E, and Γ_i are the center energy and broadening parameter, respectively, of the i-th oscillator. All energies are in eV, and the amplitude A, has a unit (cV)*. Only two oscillators were used for all three materials. The first oscillator was located at $E_i = 0$, corresponding to the free electron response (Drude model). The other was located at higher energy (≈ 6 cV), beyond our spectral range, to account for interband transitions in the UV.

S. RESULTS AND DISCUSSION

5.1 Optical constants of transparent (thin) and opaque (thick) film

Fitted oscillator parameters are given in Table 1 for thick (TK) and thin (TN) films of each material. All oscillator parameters, except E_1 which was fixed to O cV, were fixed to the values found from the wavelength-by-wavelength fitting combining VASE and transmittance data,

| Table 1. T | >-oscillator fitte | ed pa ameters. |
|------------|--------------------|----------------|
|------------|--------------------|----------------|

| 1000 1. 11 | 0001111111 | Jan Level Jack | ATTECO DI | | | |
|---|------------|----------------|-----------|------|------|------|
| | HfN | | ZrN | | TiN | |
| | TK | TN | TK | TN | TK | TN |
| $\mathbf{\epsilon}^{\infty}$ | 2.74 | 3.11 | 2,31 | 2.42 | 0.79 | 1.89 |
| $A_1(cV)^*$ | 65 | 70 | 6 2 | 65 | 52 | 57 |
| 1 't (eV) | 0.9 | 1.20 | 0.73 | 0.95 | 0,71 | 0.83 |
| E_1 (eV) | 0 | 0 | " o | 0 | 0 | 0 |
| $A_{\stackrel{\cdot}{*},\stackrel{\cdot}{*},\stackrel{\cdot}{*}}$ | 99 | 101 | 124 | 153 | 202 | 160 |
| *************************************** | 2.8 | 3.8 | ·2.7 | 4.2 | 4.3 | 4 |
| E_2 | 6.1 | 6.2 | .6.1 | 6.6, | 6.3 | 5.8 |
| χ | 3.8 | 27 | ,2,7 | 22, | 9.8 | 15 |
| <i>t</i> (nm) | | 39.5 | | 40.7 | • | 34.1 |

Note the values of χ^2 which indicate how good the fits were. The oscillator model works well for ZrN and HfN thick films (low χ^2), but not nearly as good for thin films. It also works less well for the TiN thick film. Figures 1 and 2 show real (n) and imaginary (k) parts of the complex index of refraction for thick (l'K) TiN, ZrN, and HfN films determined by VA SE over the spectral wavelength of 240 1700 nm respectively. Notice in Figure 1 that the real part of the refractive index, n, is close to or below 1 in the major part of the visible spectrum (440-670 rim).

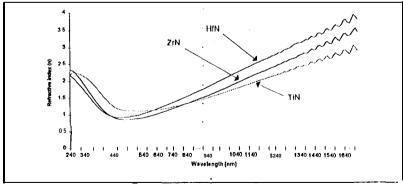
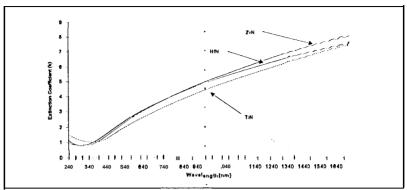


Figure 1. Real part Of the refractive index of TK • TiN, ZrN and HfN.

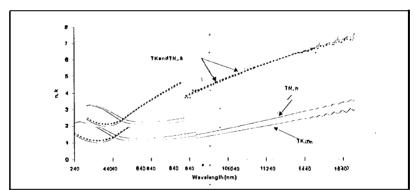
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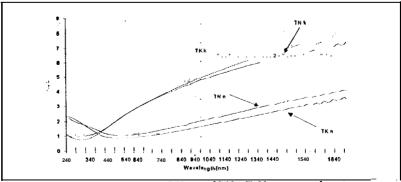


igure 2. Imaginary part of refractive index of TK • TiN, ZrN and HfN.

Figures 3, 4 and S compare n and k values for thick and thin films of TiN, ZrN and HfN respectively. Results show that optical constants for the films, prepared at the same deposition conditions, not only depend on the wavelength of light, but also on the thickness, t, of the films. TiN shows less difference between thick and thin film optical constants and the difference is distributed differently than for the other two materials. The results imply that n and k are functions of the film's thickness, at' least over some range of thickness with ZrN and HfN, and to a lesser extent with TiN. Most of the dependence on thickness is in the longer wavelength regions where free electron absorption dominates. If in particular applications all films are to be very thick (hundreds of m), then the variations of n and k in the films are minimal.



'igure 3. Optical constants of TK • and TN • TiN measured with VASE.



igure 4. Optical constants of TK • and TN • ZrN measured with VASE

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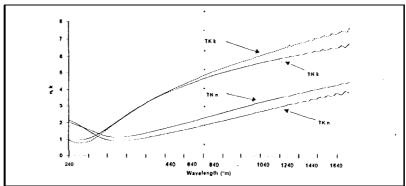
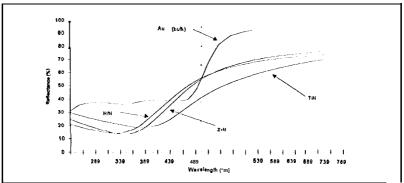


Figure 5. Optical constants of TK • and TN • HfN measured with VASE,

The spectral reflectance and transmittance of the three nitride films are calculated using a commercial film design software package where the input data are the measured n, k, and film thickness t. The films are assumed to be coated on a non-absorbing fused silica substrate with $n_s = 1.46$. Figures 6-8 present the calculated spectral reflectance and transmittance as a function of wavelength for these films. The results imply that the spectral behavior of TiN, ZrN and HfN films are similar to the noble metals Au, Ag and Pt. Reflectance curves increase slowly with increasing wavelength, approaching their maximum of ~93% in the infrared. The transmittance curves increase slowly with decreasing wavelength towards a maximum in the mid-lower side of the visible spectrum and there is a corresponding reflectance minimum at these wavelengths. If we compare the reflectance and transmittance of these three nitrides to that of noble metals, TiN is most similar to Au, and HfN and ZrN are similar to Ag.



'igure 6. Spectral reflectance vs. wavelength for thin films of TiN, ZrN, H' and bulk Au.

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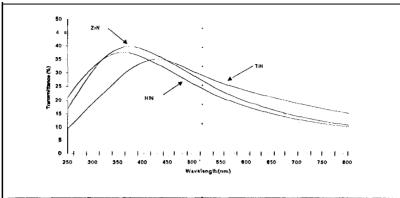


Figure 7. Spectral transmittance of thin films of TiN, ZrN and HfN.

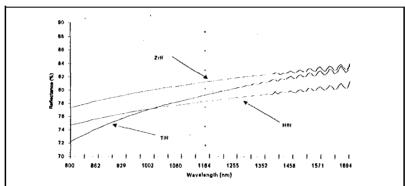


Figure 8. Calculated reflectance of TiN, ZrN and HfN thin films at . 10rmal incidence.

5.2 Composition analysis

Film composition was obtained by Rutherford backscattering analysis. The amounts of metal, nitrogen and oxygen were determined quantitatively from the areas beneath the spectra peaks and from the Rutherford cross-section of each element. The composition of the films with and without ECR is listed in 'fable 2. Oxygen is the main impurity in all films. It may be assumed that this is due to the residual pressure of oxygen in the chamber. The ratio of nitrogen to oxygen increases with the atomic weight of the starting material, HfN, with the heaviest atomic weight, has the highest nitrogen-to-oxygen atomic ratio.

5.3 Optical Scatter Characterization

Two deposit ions were made for each material -- one with and one without ECR --- onto super-polished fused silica substrates. We examined scattering from each coating using an angle-resolved optical scatterometer at the University of New Mexico. As shown in Table 3, TiN and HfN coat ings have rms roughness simi Iar to super-polished substrates; ZrN has much higher rms. One reason for this could be due to its crystalinity.

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Table 2. RBS results of 2 Mev He⁺ ions on TiN, ZrN, and HfN films on graphite substrate.

| - | | | l'ilm Composition (at%) | | | | | |
|-------------|-----------|-------|-------------------------|----|-----|-----|------|-----|
| Film | Thickness | (Å) T | i | N | O | | H | Ar |
| TiN | 48 | 4 | 8 | 48 | 4 | 0 | 0 | 0 |
| TiN | 962 | 4 | 8. | 48 | 4 | 0 | 0 | 0 |
| TiN w/ECR | 916 | 4 | 8 | 48 | 4 | 0 | 0 | 0 |
| ZrN | 102 | | 0 . | 42 | 15 | 43 | 0 | 0 |
| ZrN | 140 | | 0 . | 49 | 8 | 43 | 0 | 0 |
| ZrN | 386 | | 0 . | 52 | 4 | 44 | 0 | 0 |
| ZrN w/ECR | 53 | | 0 · | 51 | 6 | 43 | 0 | 0 |
| ZrN w/ECR | 235 | | 0, | 50 | 5 | 45 | 0 | 0 |
| ZrN w/ECR | 430 | | 0 | 49 | 4 | 47 | 0 | 0 |
| HIN | 82 | | 0 | 53 | 4,7 | 4.4 | 36 | 1.9 |
| 1 IfN | 344 | | о" | 51 | 4.6 | 4.5 | 38,4 | 1,5 |
| 1 IfN | 1335 | | o " | 52 | 4 | 4 | 40 | 0 |
| HfN w/ECR | 128 | | 0 · | 48 | 5 | 4.8 | 40.5 | 0.7 |
| 1 IfN w/ECR | 450 | | 0. | 47 | 2.5 | 4.5 | 44 | 1 |

Table 3. Rms roughness values obtained using an optical scatterometer.

| | rms | Roughness (| Å) | | |
|-----------|------|-------------|------|---------|-----------|
| | T | rial Number | | | Standard |
| | 1 | 2 . | 3 | Average | Deviation |
| 1 iN | 26.4 | 14.0, | 12 | 17.5 | 7.8 |
| TiN w/ECR | 10.6 | 7.4 | | 9.0 | 2.3 |
| ZrN | 27.5 | 76.3 | 27.9 | 43.9 | 28.1 |
| ZrN w/ECR | 23.0 | 35,1 | 46.1 | 34,7 | 11.6 |
| HfN | 14.7 | 9.1 | " - | 11.9 | 4 |
| HfN w/ECR | 25.3 | 25.0 · | - | 25.2 | 0.2 |

5.4 Electrical resistivity

Measured sheet resistance (R_x) of the films as a function of nitrogen gas flow rate during the deposition are listed in 'l'able 4. Film resistivity calculated from R_x and film thickness, t, as a function of N_2 flow rate are plotted in Figures 9-11.

Table 4. Measured sheet resistance R_s of the nitride films.

| N ₂ Flow (sccm) | R (Ω) - TiN | ,R | (Ω) – ZrN | R (Ω) – HfN |
|----------------------------|------------------------|----|--------------------|-------------|
| 3 | 1 1 . | 7 | 15 | 15 |
| 5 | 5 | | 15 | 9.9 |
| 10 | 4 . | 4 | 20 | 28 |
| 15 | 5.3 | | 30 | 235 |
| 20 | 6 | • | 99 | 1100 |
| 25 | 7 . | 7 | 120 | 14400 |
| | 9 | , | 510 | too high |

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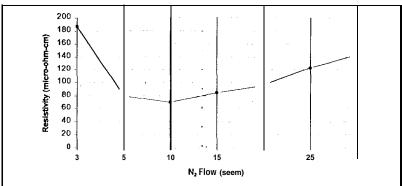


Figure 9. Room temperature resistivity of TiN as a function of nitrogen flow rate.

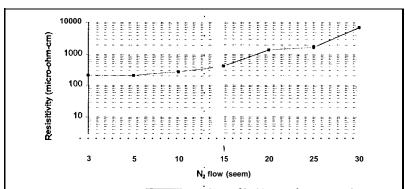


Figure 10. Room temperature resistivity of ZrN as a function of nitrogen flow rate.

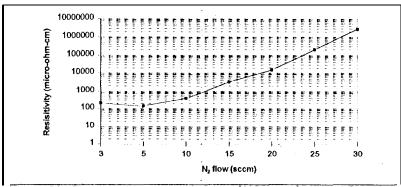


Figure 11. Room temperature resistivity of HfN as a function of nitrogen flow rate.

It is important to notice that there exists a critical nitrogen flow where the resistivity changes abruptly for ZrN and HfN, but has relatively much less effect in the case of TiN in our investigation we found that there is a link between the color, resistivity and optical reflectance of the three films. For TiN, a nitrogen-to-titanium ratio of unity results in maximum reflectance with a gold color and minimum room temperature resistivity of 75 $\mu\Omega$ -cm. For ZrN the atomic ratio of nitrogen to Zr is about 1,12 and for 11fN the ratio is 1.30, resulting in 200 and 140 $\mu\Omega$ -cm room temperature resistivity,

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respectively. By increasing or decreasing the N₂ flow the color changes to gray, optical reflectance drops, and resistivity increases in all three materials.

6. Durability

Deposited coatings of TiN, ZrN and HfN were subjected "to severe abrasion by rubbing the coated surfaces with a standard eraser conform ing to MI L-E-12397 mounted in an eraser abrasion tester. ('I'his is a US Army specificat ion for hard coatings.) For the accelerated aging test the samples were put into an environmental controlled test chamber and exposed to a temperature of 65°C and 95–) 00% relative humidity for 120 hours. To test the thermal stability of the coatings, the TiN samples were baked in an oven for three hours at 300°C in air at atmospheric pressure. The coatings did not exhibit any evidence of scratches, flaking, peeling or blistering. I lowever, the baked coating of TiN showed an approximate 30% reflectance drop in ihc 1.5 μ m to 3.0 μ m optical range. '1'here was no reflectance drop in either the 0.2–1.5 μ m or 3.0-32.0 μ m optical ranges.

7. CONCLUSIONS

Reactive DC magnetron sputtering was used to prepare thin films of TiN, ZrN and HfN at room temperature. The films exhibit optical properties similar to noble metals. The ellipsometry data implies that the n and k of the films are a function of the film's thickness - - at least over some range of thickness for ZrN and HfN, and to a lesser extent with TiN, The composition of TiN, ZrN and HfN and their corresponding optical and electrical properties can be varied by controlling nitrogen partial pressure during deposition. The room temperature resistivity values obtained in this investigation for TiN, ZrN and HfN thin films are 75 $\mu\Omega$ -cm, 200 $\Omega\mu$ -cm, and 140 $\mu\Omega$ -cm respectively. The process has the potential to produce TiN, ZrN and HfN with rms roughness comparable to that of super-polished substrates, i.e., 1- 5Å rms roughness. Film adhesion to fused silica, optical glass and silicon substrates is excellent. TiN films exhibit optical properties most similar to Au, while ZrN and HfN films are similar to Ag. Comparing the two groups, noble metals and the transition metal nitrides, we conclude that the transition metal nitrides exhibit superior mechanical and chemical properties, but not as good optical nor electrical properties.

8. ACKNOWLEDGMENTS

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